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SYNTHESIS OF DL-MYO-INOSITOL 1,4,5-TRISPHOSPHATE

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ABSTRACT: DL-Myo-inositol 1,4,5-trisphosphate has been synthesised from (\pm) 1,2,4-tri-O-benzyl-myo-inositol using a phosphite chemistry approach.

The identification of a phosphate ester of <u>myo</u>-inositol (1), D-<u>myo</u> inositol 1,4,5-trisphosphate (2), as an intracellular second messenger for calcium mobilisation¹ has regenerated a great biological interest in the inositol phosphates.² Isolation of (2) from natural sources is not practicable for large quantities of material and there is consequently a need for chemically synthesised material as well as analogues of (2) and radioactive derivatives.



A synthesis of (2) has recently been reported,³ but the phosphorylation and deblocking methods were not considered satisfactory. We report here a synthesis of racemic (2) starting from (\pm) 1,2,4-tri-O-benzyl-myo-inositol⁴ using a phosphite chemistry approach⁵ based upon di-isopropylamino(2-cyanoethyl)chlorophosphine (3),⁶ a reagent developed for DNA synthesis.

(±)-1,2,4-Tri-O-benzyl-myo-inositol (0.20 mmole) (4) was treated with (3) (0.66 mmole) and diisopropylethylamine (0.60 mmole) in dichloromethane (3 ml) to yield the trisphosphoroamidite







⊥ (iii)





All compounds are racemic Reagents: (i) ClP(OCH₂CH₂CN)NPrⁱ₂, 3.3 equiv., C₂H₅N(Prⁱ)₂, 3 equiv./CH₂Cl₂; (ii) tetrazole-HOCH₂CH₂CN, 3 equiv./CH₂Cl₂; (iii) <u>t</u>-BuOOH; (iv) Na-liq. NH₃.

(5) [δ_{p} , 150.46] in <u>ca.</u> 90% yield. A small peak at δ 146.0 due to the sole other product was tentatively assigned to material possessing a 4,5-cyclic phosphoroamidite system arising from a small amount of mono-phosphitylation of the vic diol system. Reaction of (5) with tetrazole (0.66 mmole) and 2-cyanoethanol (0.66 mmole) yielded the trisphosphite triester (6) [δ_{p} , 139.6, P-1; 140.2, 141.0, P-4, P-5 (not assigned), ${}^{5}J_{pp} = 3.4 \text{ Hz}$]. ${}^{31}P$ NMR spectroscopy showed ${}^{5}J_{pp}$ spin-spin coupling in the form of an AB system for two of the phosphite triesters, confirming bis-phosphitylation of the vicinal 4,5-diol system. Oxidation of (6) with anhydrous t-butyl hydroperoxide (3 ml) afforded quantitatively the trisphosphate triesters (7) $[\delta_{D}, -3.4]$ and finally complete deblocking with sodium in liquid ammonia⁸ effected reductive removal of the benzyl groups and removal of the 2-cyanoethyl groups by β -elimination to give crude (2), which was purified by ion exchange chromatography.⁹ Since the resulting material still contained minor amounts of inositol phosphate impurities, a small sample was subjected to final purification by HPLC.¹⁰ We also found that (2) could be purified in quantity by gradient elution using a DEAE Sephadex ion-exchanger.¹¹ DL-myo-inositol 1,4,5-trisphosphate was obtained in <u>ca</u>. 50% yield, as determined by quantitative phosphate analysis. The 31 P NMR spectrum of (2) 12 was in accord with that reported recently for material of natural origin.¹³



24.15 MHz ³¹P NMR spectrum of synthetic DL-myo-inositol 1,4,5trisphosphate (ca. 240 mM in D_2O , pH 12 with added EDTA); NMR parameters were: sweep width, 1K Hz; pulse width, 8 μ s; collected in 4K; no. of transients, 100; broad band proton decoupling; determined using a Jeol FX 60 NMR spectrometer; reference, external 85% H_2PO_4 .

We have demonstrated our synthetic route using racemic (4), however, methods have been recently reported whereby resolved material can be obtained^{3,14} and the present synthetic method can therefore be used to prepare optically active (2). Synthesis of other inositol phosphates using this methodology as well as biological testing¹⁵ are in progress.

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- 10. HPLC purification was performed on a Partial SAX 10 column using ammonium formatephosphoric acid buffer as reported: I.R. Batty, S_RR. Nahorski, and R. Irvine, <u>Biochem. J. 232</u>, 211 (1985). A small sample of [³H]-D-myo-inositol 1,4,5-trisphosphate was added to permit detection.
- 11. <u>Ca.</u> 300 µmole of (2) were purified on a column of DEAE Sephadex A-25 anion exchanger equilibrated in 100 mM triethylammonium bicarbonate (TEAB). When this column was run using a linear gradient of TEAB (0.1-1.0 M) clean separation of (2) from other phosphate-containing impurities was observed. (2) Was eluted at a buffer concentration of <u>ca.</u> 700 mM. TEAB was removed by evaporation <u>in vacuo</u> to yield the triethylammonium salt of (2) as a colourless glass.
- 12. ${}^{31}_{P}$ NMR (D₂O, pH 12) ammonium salt: δp , 3.38, J_{PH} = 5.83; 5.11, J_{PH} = 6.93 Hz; 5.21, J_{PH} = 7.60 Hz; relative to H₃PO₄; determined on a Bruker WH400 at 162 MHz.
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